SYLLABUS
CHEM 223 – Organic Chemistry A – 1st Semester
Fall 2010 - LOYOLA UNIVERSITY CHICAGO (LUC)

Lecture:  #1236  Section: 004  Tues+Thur  10:00 am – 11:15 am  Flanner – Auditorium 133

Discussion:  #1238  Section: 005  Tues  2:30 pm – 3:45 pm  Flanner 007
#1240  Section: 006  Thur  2:30 pm – 3:45 pm  Flanner 007

Sr. Lecturer:  Dr. C. Szpunar
Office:  Flanner Hall 213  Contact:  773-508-3128, cszpuna@luc.edu
Student Office Hours:  Mon, Tues, and Wed: 11:45 - 1:30 p.m. *and by prior appt


Suggested / Recommended Materials:
2.  Molecular modeling kit, Darling, Prentice-Hall, Freeman (Maruzen), Proteus, or equivalent
3.  Spiral or bound notebook for homework problems

Optional Materials (found helpful by some students):
   (Pls see instructor before purchasing, only for students who may require an alternative study approach.)

Grading (approx weight below) with grade guidelines:  > 90% A; 75-90% B; 55-75% C; grading may be curved

EXAMS – 2 – dates scheduled and announced (subject to change, although unlikely) – NO MAKE UPS 40%
•  UNEXCUSED ABSENCES merit a zero score.
•  EXCUSED ABSENCES are handled on a case-by-case basis; grade weighting may be adjusted, depending
  on the circumstance(s); however, an excused absence MUST BE CORROBORATED and
  DOCUMENTED, e.g., accompanied by a note from the doctor, dentist, hospital rep, or funeral director;
  by a court summons, plane ticket stub, hospital release form, obit, or other. With proper documentation,
  religious observance, representing the university, or personal emergency constitutes an Excused Absence.
QUIZZES – TBD – unannounced (during lecture or discussion or as take-home, no make-ups) 20%
HOMEWORK - assigned per topic, see Suggested Homework Assignment, below 5%
FINAL EXAM – date scheduled and announced (scheduled by CAS) 35%

Course Objective: To guide, encourage, and foster the learning and understanding of Organic Chemistry –
nomenclature, structures, properties, reactions, mechanisms, and syntheses – by the individual student, helping him/her
to connect, extrapolate, integrate, and apply the many different aspects learned.

Student Outcomes: The successful student will learn how to …
1.  identify the various classes of organic compounds, their methods of preparation, and typical reactions.
2.  name and draw specific organic compounds.
3.  postulate a logical reaction mechanism for simple organic reactions.
4.  discriminate amongst relative stabilities of reaction intermediates.
5.  plan and write out multi-step syntheses using known functional group transformations.
6.  prepare for basic purification/separation techniques of organic compounds required in the laboratory.
7.  analyze and interpret data from various instruments used in separating and identifying organic
   compounds: IR, NMR, and UV-vis spectrophotometers and mass spectrograph.
Lecture and Discussion – Attendance and Attention: Important and required. Feel free to bring your books and modeling kit to class. Better yet, use them. Prepare for lecture by prior scanning of new material. Come prepared for discussion, ready to ask questions on assigned homework or yet unassimilated lecture material.

Phones and Pagers: Please be courteous and respectful of others. Silent mode during lecture and discussion. Not allowed in sight or within hearing during exams, subject to confiscation. NO phone conversations in lecture hall or in discussion class – before class, during class, after class – AT ANY TIME!

Academic Honesty: Essential, expected, and enforced. Dishonesty dictates consequences which may include: (1) notification of Chemistry Department Chair, student’s Department Chair, and CAS Dean, (2) documentation in the student’s official university record, and (3) dismissal from the university. Immediate consequences will include a ZERO on any item in question (quiz or exam). Please refer to the LUC Undergraduate Handbook on policies or the CAS website: http://www.luc.edu/cas/pdfs/CAS_Academic_Integrity_Statement_December_07.pdf.

Study Strategies and Suggestions: One may approach the study of Organic Chemistry in a manner similar to tackling a new foreign language. Its study will provide a basis to understanding future material – building constantly, incessantly, and relentlessly on the structural and mechanistic information presented previously. Over two semesters, the course will cover functional groups, aliphatic and aromatic compounds, bonding, nomenclature, stereochemistry, conformational analysis, reaction mechanisms, multi-step syntheses, and spectroscopy. Because the course is cumulative and builds heavily on prior material, the best plan is to study Organic Chemistry regularly, every day, similar to practicing the piano. Collaboration on homework problems is encouraged, especially in a timely fashion. Experience dictates that positive outcomes (for exam and course grades) are directly proportional to working and understanding the assigned problems on a regular basis, i.e., applying the concepts learned to non-generic compounds.

Typically, Organic Chemistry is not efficiently self-taught. Overnight cramming will probably not produce success. The student should quickly read the chapter/segment to be covered BEFORE lecture to improve lecture comprehension. After lecture, careful detailed re-reading of the chapter/segment and focused working of the assigned problems are appropriate, necessary, and expected. In addition to student’s participation in lecture, discussion, reading, and homework, joining and contributing to a study group is encouraged. If anticipating a passing grade of C, the minimal time per week devoted to Organic Chemistry is estimated at 4 hr for lecture and discussion, 4-10 hr for reading, and 4-10 hr for homework.

Suggested Homework Assignment (for Wade’s 7th edition):

Chap 2: 3-5, 7-11, 15-23, 35-36, 39-42, 44
Chap 3: 1-5, 9-10, 14-17, 20-21, 24-25, 29-30, 32-35, 37, 39, 44, 46
Chap 4: 1-2, 4-13, 18-22, 26, 28-31, 34-39, 42, 45-46, 50
Chap 5: 1-7, 15-22, 25-31
Chap 6: 1-7, 9-12, 14-16, 19-20, 22-24, 30-38, 40-45, 53, 56
Chap 7: 1-2, 4-8, 11, 13, 15, 19, 21, 23-25, 30-36, 38, 40-46
Chap 8: 1-2, 4-15, 17-19, 21-22, 27-29, 32, 34-37, 46-47, 49, 63-64
Chap 9: 1-2, 5-8, 12, 18-19, 23-29, 33-34, 36-37
Chap 10: 1-4, 7-10, 13-20, 23-26, 30-33, 37-39, 49, 51
Chap 11: 1-2, 5-6, 9-13, 21-22, 26, 31, 34, 39-45
Chap 12: 2-7, 11-12, 14-17, 23, 25
Chap 13: 2-11, 14-15, 22-25, 32-36, 38-44
Chap 14: TBD by student

Homework Package: *** due Tuesday, Nov. 16, 2010 by noon ***

Hand in at least 15 completed problems (or parts of multiple-reaction problems) per chapter, Chapters 1–10. (Underlined problems are particularly insightful or anticipatory.) If not using a notebook, include your name and the date worked on each page. Mark the CHAPTER and problem number for full credit. For ease in grading, staple package in chapter order ***and*** flag chapters in homework notebook (which will be returned to the student).
(for Wade 6th edition):

Chap 2: 5, 7-11, 15-23, 35-36, 39-42, 44
Chap 3: 1-5, 9-10, 14-15, 20-21, 29-35, 37, 39, 44, 46
Chap 4: 1-2, 4-13, 18-22, 26, 28-31, 34-39, 42, 45-46, 50
Chap 5: 1-3, 5-8, 16-22, 25-30
Chap 6: 1-7, 11-12, 14-16, 19-20, 22-24, 31-37, 44-45, 53, 56
Chap 7: 1-2, 4-5, 7-8, 13, 15, 19, 21, 23, 25, 34-35, 38, 40, 42-45
Chap 8: 1-2, 4-6, 8-15, 17-19, 27-28, 32, 34-37, 47, 49, 63-64
Chap 9: 1, 2, 5-8, 12, 18-19, 23-24, 33-34
Chap 10: 3, 7, 8-10, 13-15, 17-20, 24-26, 37-39, 49
Chap 11: 2, 5-6, 9-11, 13, 22, 31, 34, 41-45
Chap 12: 2-12, 14-19, 23, 25
Chap 13: 2-36, 38-45
Chap 14: TBD
## Lecture Outline (tentative, subject to change)

<table>
<thead>
<tr>
<th>Week</th>
<th>Date</th>
<th>Chapter</th>
<th>Topic</th>
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<tbody>
<tr>
<td>1</td>
<td>Aug 31</td>
<td>1-1</td>
<td>Intro: Lewis structures, bonding, resonance, acid-base, nomenclature...</td>
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<td>Sept 2</td>
<td>1-2 / 2-1</td>
<td>Structure and properties</td>
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<td>2</td>
<td>Sept 6</td>
<td>***</td>
<td><strong>MONDAY</strong>leftrightarrow Labor Day – Holiday** ***</td>
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<td>Sept 7</td>
<td>2-2</td>
<td>Alkanes, cycloalkanes, bicyclics</td>
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<td>Sept 9</td>
<td>3-1</td>
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<td>3</td>
<td>Sept 14</td>
<td>3-2</td>
<td>Chemical rxns - free radical halogenation, kinetics, intermediates...</td>
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<td>Sept 16</td>
<td>4-1</td>
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<td>Sept 21</td>
<td>4-2</td>
<td>Stereochemistry – chirality, isomers ...</td>
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<td>Sept 23</td>
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<td>5</td>
<td>Sept 28</td>
<td>5-2</td>
<td>Alkyl halides, nucleophilic substitution and elimination</td>
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<td>Sept 30</td>
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<td>6</td>
<td>Oct 5</td>
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<td><strong>TUESDAY</strong>leftrightarrow EXAM I (Chapters 1-5)** ***</td>
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<td>Oct 7</td>
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<td>7</td>
<td>Oct 12</td>
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<td><strong>TUESDAY</strong>leftrightarrow MIDTERM BREAK ** ***</td>
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<td>Nov 5</td>
<td>***</td>
<td><strong>FRIDAY</strong>leftrightarrow (last day to withdraw with a W) ** ***</td>
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<td>Nov 9</td>
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<td><strong>TUESDAY</strong>leftrightarrow EXAM II (Chapters 6-9)** ***</td>
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<td>Nov 16</td>
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<td><strong>TUESDAY</strong>leftrightarrow HW assignment due for grade</td>
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<td>Alcohols – rxns</td>
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<td>Nov 23</td>
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<td><strong>THANKSGIVING DAY – HOLIDAY</strong> ***</td>
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<td>Nov 25</td>
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<td>14</td>
<td>Nov 30</td>
<td>12-1</td>
<td>Spectroscopy – IR and MS</td>
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<td>Dec 2</td>
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<td>Spectroscopy – NMR</td>
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<td>15</td>
<td>Dec 7</td>
<td>13-2</td>
<td>Ethers, epoxides, sulfides</td>
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<td>14-1</td>
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<td>16</td>
<td>Dec 14</td>
<td>Tues</td>
<td><strong>Cumulative FINAL EXAM, 1:00 p.m. – 3:00 p.m., FH-133</strong></td>
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