Chemistry 102

Course: Chemistry 102 Date: Saturdays Time: 8:30A-12:20P Location: Cudahy Hall 207 Textbook: Kotz, Treichel and Weaver: Chemistry and Chemical Reactivity (7th ed.) Website: Blackboard (blackboard.luc.edu) Instructor: Prof. Jacob Ciszek Flanner Hall 122 Phone: (773) 508-3107 E-mail: jciszek@luc.edu Note, contact is best through e-mail. Don't contact me through the OWL system. I check it infrequently.

Course Philosophy: While every student may not enter the class with the intent to become proficient in chemistry, I strongly feel that my role as a professor is to get you to that point. Thus the expectations of you the student is through hard work, attending class, and completion of the homework you will obtain this proficiency and do well in the course. My role is to provide you with the information and the tools, in a coherent matter, so that solving said homework (as well as the quizzes and exams) are not burdensome.

We have a limited amount of days allotted to class. Thus it is very important that the class environment is free of distractions. No laptops or other computers are allowed. Cell phone use including texting is not acceptable.

<u>Office Hours:</u> Office hours are formally announced the first day of class. Office hours consist of one hour during each of the following time slots (3h total):

Weekday, 1h AM, (*TBA*) Weekday, 1h evening, (*TBA*) Weekend, 1h, (*TBA*)

Academic Honesty & Discipline: Honesty is the foundation of the academic system and hence is of the utmost importance. All exam and quiz answers should be exclusively your own work and no outside materials are allowed. In the unfortunate event that a student is caught cheating, 100 points will be deducted from your total grade and you will be brought to the attention of the Department Chair and Dean of the College who will determine if further action should be taken.

<u>Grading</u>: As we only have 13 lectures, this time must be used judiciously. Every week multiple homework assignments are due. Exams will be approximately every 4 weeks. Every other week will have a quiz.

The role of homework is to refresh the lecture information in your mind and prepare you for quizzes and exams. Homework will primarily be posted on OWL but expect occasional paper assignments. Homework will be due Tuesday (OWL), Thursday (OWL), and Saturday (OWL+paper). Collaboration on homework is allowed. OWL does not mimic the style of questions on the exam/quizes, so do not use it for exam prep; its main use is as a quick review of the lecture material and to see which concepts you're having trouble with.

Quizzes are designed such that an average student who works hard and grasps the material should score ~85%. Exams (and the final) are designed so that this same student will score approximately 70-75%. This allows motivated students to truly go beyond what is expected and to distinguish themselves, not to penalize those who work as expected.

<u>There are no makeup exams or quizzes</u>. Both the lowest quiz and exam are dropped. This should be used judiciously. Note that exam 4 will be take-home so we can get it in before the end of the school year. You will have 24h from the time when the exam 4 is available for pickup/download to complete and return the exam.

Discussion points are given for the final portion of the class where you are expected to work through some selected problems in small groups. The purpose here is to work through material presented in lecture.

Finally, there is a grading category called office hours. You are expected to stop by at least one office hour, within the first four weeks (by 10/3), even if it is just to introduce yourself. This will be the easiest 10 points you earn.

Grading scale:

| 12×10 pts | 120 | A>90% |
|--------------------|--|---|
| 5×20 pts | 100 (lowest of 6 dropped) | B> 80% |
| 3×100 pts | 300 (lowest of 4 dropped) | C>70% |
| 200 pts | 200 | D>60% |
| 12×5 pts | 60 | |
| 10 pts | <u>10</u> | |
| - | 790 | |
| | $5 \times 20 \text{ pts}$ $3 \times 100 \text{ pts}$ 200 pts $12 \times 5 \text{ pts}$ | $5 \times 20 \text{ pts}$ 100 (lowest of 6 dropped) $3 \times 100 \text{ pts}$ 300 (lowest of 4 dropped)200 pts200 $12 \times 5 \text{ pts}$ 6010 pts10 |

Note, the intended scale for exams and the final would put the average just above the lowest C. Homework and quizzes will mitigate this a bit. Based on overall class competence the grading scale <u>may</u> be relaxed a little at the end of the semester (certainly no more than a couple percent). If the whole class has mastered the material, you will not be punished because you are below the average. The A, B, C, D scale represents the maximum score you would need for that grade. Pluses and minuses are not indicated in the grading scale but will be given. This will be done according to the natural breakdown of the grade distributions. Expect this to be the closest 1-2% to the final A-B, B-C, and C-D divisions.

| Bullu | Scheune (including approximate page numbers). | | | | | |
|-------|--|------------|---------------|--------|--|--|
| 9/3 | Labor Day Weekend, no class | | | | | |
| 9/10 | Introduction, Solutions | p617-648 | | | | |
| 9/17 | Kinetics | p670-712 | HW#1 due TTS | Quiz 1 | | |
| 9/24 | Exam 1, Equilibrium | p724-743 | HW#2 due TTS | Exam 1 | | |
| 10/1 | Equilibrium, Acids & Bases | p744-787 | HW#3 due TTS | Quiz 2 | | |
| 10/8 | Acid & Bases, Aqueous Equilibrium | p788-821 | HW#4 due TTS | | | |
| 10/15 | Aqueous Equilibrium | p822-850 | HW#5 due TTS | Quiz 3 | | |
| 10/22 | Exam 2, Thermodynamics | p860-867 | HW#6 due TTS | Exam 2 | | |
| 10/29 | Thermodynamics, Electrochemistry | p867-886 | HW#7 due TTS | Quiz 4 | | |
| 11/5 | Electrochemistry | p896-940 | HW#8 due TTS | | | |
| 11/12 | Electrochemistry, Coordination Chem. | p1018-1028 | HW#9 due TTS | Quiz 5 | | |
| 11/19 | Exam 3, Coordination Chemistry | p1029-1053 | HW#10 due TTS | Exam3 | | |
| 11/26 | Thanksgiving Break, no class | | | | | |
| 12/3 | Nuclear Chemistry | p1060-1090 | HW#11 due TTS | Quiz 6 | | |
| 12/4 | Exam 4 available 12/4 8p (on blackboard), due 12/5 8p | | | | | |
| 12/10 | Nuclear Chemistry, Review***, no discussion pts | | HW#12 due TT | | | |
| 12/19 | Final 4:15p | | | | | |
| | Note the near in the first two weeks will be measure. Do managed | | | | | |

Schedule (including approximate page numbers):

Note, the pace in the first two weeks will be rigorous. Be prepared.

Midsemester break (10/10, 10/11) does not include our course

Other:

Simple calculators are allowed for exams. Those capable of storing complex (for example images) or large amounts (1+ pages of text) of information are not. See me if you are unsure about yours.

The Tutoring Center offers free small group tutoring and lab (drop-in) tutoring for Loyola students. The groups meet once a week through the end of the semester and are led by a student who has successfully completed study in the course material. To learn more or request tutoring services, visit the Tutoring Center online at www.luc.edu/tutoring.